# Pyranonaphthoquinone Antibiotics. Part 1. Syntheses of 9-Demethoxyeleutherins and 9-Deoxynanaomycin A Methyl Ester 


#### Abstract

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The syntheses of 9-demethoxyeleutherins and 9-deoxynanaomycin A methyl ester starting from indan-1-one derivatives are described. Lemieux-Johnson oxidation of the indene (15) derived from 4,7-dimethoxy-2-methyl-indan- 1 -one (14) afforded the diketone (16). The diol (17) obtained by lithium aluminium hydride reduction of (16) was treated with hydrochloric acid to give a ca. $1: 2$ mixture of cis-5,8-dimethoxy-1,3-dimethylisochroman (18) and the trans-isomer (19). Treatment of (16) with hydrogen bromide in acetic acid followed by catalytic reduction gave exclusively the cis-isochroman (18). Oxidative demethylation of the isochromans (18) and (19) afforded the quinones (22) and (23), benzannelation of which in two steps yielded 9-demethoxyeleutherin (24) and 9 -demethoxyisoeleutherin (25) respectively. The same oxidation of the indene (32) derived from 4,7-dimethoxyindan-1-one (28) afforded the keto-aldehyde (33), which was treated with methoxycarbonylmethylenetriphenylphosphorane to give the conjugated ester (34). Reductive cyclisation of (34) with sodium borohydride afforded a ca. 1:3.5 mixture of cis-5,8-dimethoxy-3-methoxycarbonylmethyl-1-methylisochroman (35) and the trans-isomer (36). Oxidative demethylation of (36) followed by benzannelation produced 9-deoxynanomycin A methyl ester (39).


Nanaomycins A (1) ${ }^{1}$ and D(2), ${ }^{2}$ griseusin A (3), ${ }^{3}$ and granaticin (4) ${ }^{4}$ are representatives of a group of pyranonaphthoquinone antibiotics. These natural products are noteworthy for their biological effects as antimicrobial agents. The nanaomycins, in particular, were reported ${ }^{1 b}$ to have excellent activities against mycoplasmas. Synthetic efforts in this area have yielded the first total synthesis of racemic nanaomycins A and D in 1978. ${ }^{5}$ We have recently embarked on a programme exploring a new synthetic route to nanaomycins, eventually leading to a total synthesis of even more complex molecules such as griseusin A and granaticin.

(1)

(3)

(2)

(4)

Methods hitherto reported for the construction of the isochroman (or $\gamma$-lactoisochroman) ring system could be classified into two categories as illustrated in Scheme 1. The first (a) ${ }^{5,6}$ originates from Schmid's method ${ }^{7}$ for the synthesis of eleutherin and isoeleutherin. ${ }^{8}$ The key step involves condensation of the $\gamma$-naphthyl- $\beta$-hydroxyester (5) with acetaldehyde giving the isochroman (6), which could be converted into the quinone (7), and
further into the $\gamma$-lactone (8), by air oxidation. The second approach (b) by Kraus ${ }^{9}$ features the Michael addition of a butenolide anion equivalent, 2 -t-butoxyfuran, with the acetylnaphthoquinone (9). The condensation product (10) was then reductively cyclised to
(a)

(5)
(6)



(11)

(7)
(8)

(10)

(b)

(9)

Scheme 1
give the $\gamma$-lactopyran derivatives (11). We have investigated the use of readily accessible indanone derivatives as precursors of the heterocyclic ring system. Results obtained with the synthesis of 9 -demethoxyeleutherins performed as a model study and its extension to the synthesis of 9 -deoxynanomycin $A$ are now described.

Synthesis of 9-Demethoxyeleutherin (24) and 9-Demethoxyisoeleutherin (25).-For the synthesis of the starting indan-1-ones, we have used the Fries rearrangement and subsequent Friedel-Crafts reaction ${ }^{10}$ of appropriate aryl esters of unsaturated acids. Careful heating of the methacrylate (12) with aluminium chloride-sodium chloride $\left(90^{\circ} \mathrm{C}\right.$ for $40 \mathrm{~min}, 120^{\circ} \mathrm{C}$ for 50 min , and then $180{ }^{\circ} \mathrm{C}$ for 10 min ) afforded 4,7-dihydroxy-2-methyl-indan-l-one (13), which was treated with sodium hydroxide-dimethyl sulphate to give 4,7-dimethoxy2 -methylindan-1-one (14) (Scheme 2).

Reaction of the indanone (14) with methylmagnesium iodide yielded an unstable Grignard product, which, when shaken in ether with dilute hydrochloric acid, was transformed to 4,7-dimethoxy-2,3-dimethylindene (15) in $72 \%$ yield. Oxidative cleavage of the double bond of

the indene (15) by the Lemieux-Johnson method produced an oily diketone (16) in $72 \%$ yield. The crude (16) was reduced, without further purification, with lithium aluminium hydride to give the diol (17). The diol was then subjected to an acid-catalysed cyclisation.

Benkeser and Johnston ${ }^{11}$ reported that the cyclisation of an analogous diol (20) with toluene- $p$-sulphonic acid in acetic acid at $100{ }^{\circ} \mathrm{C}$ gave two products, the major being an isobenzofuran and the minor an isochroman derivative. When we attempted to cyclise the diol (17) under milder conditions [stirring an ethereal solution of (17) with hydrochloric acid], it was dehydrated readily to give only 1,3-dimethylisochroman as a ca. 1:2 mixture of the stereoisomers (18) and (19) (by g.l.c.), which could be separated by silica gel chromatography. It should be noted that no isobenzofuran derivative was detected. The isochroman structures of (18) and (19) were supported by their n.m.r. spectra, each showing two $3-\mathrm{Me}$ and 1 -Me doublets at $\delta 1.34$ and 1.54 (each $J 7 \mathrm{~Hz}$ ) for (18) and at 1.31 and 1.49 (each $J 7 \mathrm{~Hz}$ ) for (19). The stereochemistry of the isomers was assigned by comparing long-range coupling constants ( $J_{1,4}$ ). The minor product (18) which exhibited larger coupling constants ( $J_{1 a^{\prime}, 4 a^{\prime}} 2 \mathrm{~Hz} ; J_{1 \mathrm{a}^{\prime}, 4 \mathrm{e}^{\prime}} 1 \mathrm{~Hz}$ ) was assigned the cis-structure, and the major ( $J_{1.4}<1 \mathrm{~Hz}$ ) the trans-isomer. This stereochemical assignment was supported by an acid-
catalysed equilibration experiment: the cis-isomer (18) could be epimerised in methanesulphonic acid to the thermodynamically more stable trans-isomer giving a ca. 1:4 mixture of (18) and (19) (Scheme 3).


The cis-isomer was most conveniently obtained from the diketone (16) by the two-step procedure in Scheme 4. Treatment of (16) with hydrogen bromide in acetic acid produced the deeply coloured oxonium salt (21), which was characterised by two downfield singlets at $\delta 2.07$ and 2.94 corresponding to the $\mathrm{C}-1$ and $\mathrm{C}-3$ methyl groups. Catalytic hydrogenation of the salt over platinum gave exclusively the cis-isochroman (18) in an overall yield of $40 \%$ from (16), only a trace anount of the trans-isomer (19) being detected by g.l.c.


Oxidative demethylation of the isochromans (18) and (19) was then carried out with cerium(Iv) ammonium nitrate in aqueous acetonitrile to produce the corresponding quinones (22) and (23) in the same yield ( $85 \%$ ). The difference between the long-range coupling constants for $1-\mathrm{H}$ and $4-\mathrm{H}$ for the quinone isomers was quite distinctive, as found for eleutherin (26) and isoeleutherin (27): ${ }^{12} 4.0$ and 2.8 Hz for (22); 2.0 and $<1.0$ Hz for (23).

The final benzannelation was performed in two steps in excellent yields. Treatment of (22) with 1-acetoxy-
buta-1,3-diene in toluene at room temperature afforded a cycloaddition product, presumably as a mixture of two regioisomers. The crude adduct was then stirred with sodium carbonate in aqueous ethanol to afford 9-demethoxyeleutherin (24) in $89 \%$ yield. 9-Demethoxyisoeleutherin (25) was also obtained by the same procedure from (23) in $91 \%$ yield (Scheme 5).

Synthesis of 9-Deoxynanaomycin A Methyl Ester (39).We first attempted to synthesise the $\beta$-keto-ester (31) starting with the indanone (29) via the indene (30) by employing essentially the same procedure as just described. However, Lemieux-Johnson oxidation of the indene (30) did not lead to any identifiable product. Therefore, we investigated an alternative route, the key intermediate of which was the conjugated ester (34).


4,7-Dimethoxyindan-1-one (28) obtained by Buck's method ${ }^{13}$ was treated with methylmagnesium iodide and the product was briefly treated with hydrochloric acid to give 4,7-dimethoxy-3-methylindene (32) in $85 \%$ yield. The double bond of the indene (32) was then subjected to oxidative cleavage by the Lemieux-Johnson method. The resulting unstable keto-aldehyde (33) was immediately treated with a slight excess of methoxycarbonylmethylenetriphenylphosphorane in benzene at room temperature to yield the conjugated ester (34), shown to be the stereochemically homogeneous $E$-isomer by n.m.r. analysis ( $J_{v i c} 16 \mathrm{~Hz}$ ) (Scheme 6).

Reduction of (34) with sodium borohydride in methanol produced two products in a ratio of $1: 3.5$ (by g.l.c.). Both compounds, isolated by preparative t.l.c., were shown by spectral data (n.m.r. and m.s.) to possess an isochroman structure. The major, more polar product which showed smaller values of $J_{1,4}(<1 \mathrm{~Hz})$ was tentatively assigned the trans-isochroman structure (36) and the other $\left(J_{1^{a^{\prime}}, 4^{a^{\prime}}}=J_{1^{a^{\prime}, 4^{\prime}}}=1.5 \mathrm{~Hz}\right)$ the cisstructure (35). The isochromans (35) and (36) were oxidatively demethylated with cerium(rv) ammonium nitrate in aqueous acetonitrile to afford the corresponding quinones (37) and (38) respectively. The values of the long-range coupling constants $J_{1.4}$ for (37) (3.7 and 2.5

Hz ) and (38) (2.2 and $<1 \mathrm{~Hz}$ ) were compatible with the foregoing relative configurations. Compound (38) was converted into (39) by the procedure just described: Diels-Alder reaction of the quinone (38) with 1-acetoxy-

buta-1,3-diene followed by treatment with sodium carbonate formed deoxynanaomycin A methyl ester (39) in $78 \%$ yield (Scheme 6). The structure of (39) was fully confirmed by its m.p. $\left(185-187{ }^{\circ} \mathrm{C}\right.$; lit. ${ }^{6}$ $185-187^{\circ} \mathrm{C}$ ) and spectral (lata (n.m.r., i.r., and m.s.).


## EXPERIMENTAL

M.p.s were determined with a Yanagimoto micro melting point apparatus. I.r. spectra were taken on a Hitachi 215 spectrometer and n.m.r. spectra on a JEOL MH-100 spectrometer using tetramethylsilane as an internal standard. Mass spectra were measured with a Hitachi RMU6 MG or a JEOL D-300 spectrometer. T.1.c. was performed on Merck silica gel $60 \mathrm{~F}_{254}$ plates and preparative t.l.c. was carried out on Merck silica gel $60 \mathrm{PF}_{254+366}$. Merck silica gel 60 ( $70-230$ mesh) was used for column chromatography. G.l.c. analyses were performed on a Yanagimoto G-1800 apparatus.

4,7-Dihydroxy-2-methylindan-1-one (13).-A mixture of 4methoxyphenol ( $10 \mathrm{~g}, 0.08 \mathrm{~mol}$ ) and methacryl chloride $(9.3 \mathrm{~g}, 0.089 \mathrm{~mol})$ was heated at $70^{\circ} \mathrm{C}$ for 30 min and then at $100{ }^{\circ} \mathrm{C}$ for 20 min . The mixture was added to ether $(100 \mathrm{ml})$, the ethereal solution washed with 2 m -sodium hydroxide solution and dried ( $\mathrm{MgSO}_{4}$ ), and the solvent removed under reduced pressure to yield 4-methoxyphenyl methacrylate (12) ( $13.3 \mathrm{~g}, 86 \%$ ) as an off-white solid, $\nu_{\text {max }}$. (Nujol) $1720 \mathrm{~cm}^{-1}, \delta\left(\mathrm{CDCl}_{3}\right) 2.00(3 \mathrm{H}, \mathrm{s}, \mathrm{C}-\mathrm{Me}), 3.72(3 \mathrm{H}$, $\mathrm{s}, \mathrm{OMe}), 5.64 \mathrm{br}(1 \mathrm{H}, \mathrm{s}$, olefinic H$), 6.24 \mathrm{br}(1 \mathrm{H}, \mathrm{s}$, olefinic $\mathrm{H})$, and $6.72-7.04(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), \mathrm{m} / \mathrm{e} 192\left(M^{+}\right), 164$ ( $M^{+}-\mathrm{CO}$ ), and $124\left(\mathrm{C}_{7} \mathrm{H}_{8} \mathrm{O}_{2}{ }^{+}, 100 \%\right)$. A mixture of the methacrylate (12) ( $13.3 \mathrm{~g}, 0.069 \mathrm{~mol}$ ), aluminium chloride ( $66 \mathrm{~g}, 0.5 \mathrm{~mol}$ ), and sodium chloride $(26.5 \mathrm{~g}, 0.45 \mathrm{~mol}$ ) was heated at $90^{\circ} \mathrm{C}$ for 40 min , at $120^{\circ} \mathrm{C}$ for 50 min , and then at $180{ }^{\circ} \mathrm{C}$ for 10 min under nitrogen. The mixture was added to water, the resulting suspension extracted with ethyl acetate, the organic layer washed with $10 \%$-hydrochloric acid and dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent evaporated off to yield a solid. Crystallisation from chloroform afforded 4,7-dihydroxy-2-methylindan-1-one (13) (5.7 g, $46 \%$ ), m.p. $160-163{ }^{\circ} \mathrm{C}$; $\nu_{\text {max. }}(\mathrm{KBr}) 3230$ and $1645 \mathrm{~cm}^{-1}$; $m / e 178\left(M^{+}, 100 \%\right)$ and $163\left(M^{+}-\mathrm{Me}\right)$ (Found: C, $67.2 ; \mathrm{H}, 5.9 . \quad \mathrm{C}_{10} \mathrm{H}_{10} \mathrm{O}_{3}$ requires $\left.\mathrm{C}, 67.4 ; \mathrm{H}, 5.7 \%\right)$.

4,7-Dimethoxy-2-methylindan-1-one (14).-To a solution of the indanone (13) ( $10 \mathrm{~g}, 0.056 \mathrm{~mol}$ ) in 2 m -sodium hydroxide solution ( 60 ml ) dimethyl sulphate was added dropwise ( $20 \mathrm{~g}, 0.16 \mathrm{~mol}$ ) over 30 min and the mixture was stirred at room temperature for 30 min and then heated at $70^{\circ} \mathrm{C}$ for 10 min . The resulting mixture was extracted with ethyl acetate, the extract washed with $10 \%$ sodium hydroxide solution and water, and dried $\left(\mathrm{MgSO}_{4}\right)$. The solvent was evaporated off under reduced pressure to give a solid. Repeated crystallisation from ethyl acetate-ether afforded 4,7-dimethoxy-2-methylindan-1-one (14) ( $8.6 \mathrm{~g}, 75 \%$ ), m.p. $79.5-82.5{ }^{\circ} \mathrm{C}$; $v_{\max .}(\mathrm{KBr}) 1705 \mathrm{~cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right) 1.25(3 \mathrm{H}$, $\mathrm{d}, J 7 \mathrm{~Hz}, 2-\mathrm{Me}), 2.40-3.40(3 \mathrm{H}, \mathrm{m}, 2$ - and $3-\mathrm{H}), 3.79$ and 3.83 (each $3 \mathrm{H}, \mathrm{s}, 4$ - and $7-\mathrm{OMe}$ ), $6.64(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}$, Ar$\mathrm{H})$, and $6.90(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}) ; m / e 206\left(M^{+}, 100 \%\right)$, $191\left(M^{+}-\mathrm{Me}\right)$, and $177\left(\mathrm{C}_{10} \mathrm{H}_{0} \mathrm{O}_{3}{ }^{+}\right)$(Found: C, 69.6; $\mathrm{H}, 6.6 . \quad \mathrm{C}_{12} \mathrm{H}_{14} \mathrm{O}_{3}$ requires $\mathrm{C}, 69.9 ; \mathrm{H}, 6.8 \%$ ).

4,7-Dimethoxy-2,3-dimethylindene (15).-An ethereal solution of methylmagnesium iodide ( 2 m -solution; 10 ml , 0.02 mol ) was added dropwise, over 5 min , to a solution of the indanone (14) ( $2.0 \mathrm{~g}, 9.7 \mathrm{mmol}$ ) in ether ( 150 ml ) at $0{ }^{\circ} \mathrm{C}$ and the mixture was stirred at room temperature for 50 min , poured into saturated ammonium chloride solution, and extracted with ether. The ethereal extract was vigorously shaken with $10 \%$ hydrochloric acid for 2 min , washed with dilute sodium hydrogen carbonate solution, dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent removed under reduced pressure. Column chromatography on silica gel, with
benzene as eluant, afforded 4,7-dimethoxy-2,3-dimethylindene (15) ( $1.78 \mathrm{~g}, 72 \%$ ), m.p. $53-55{ }^{\circ} \mathrm{C}$ (from $\mathrm{Et}_{2} \mathrm{O}$-hexane); $\delta\left(\mathrm{CDCl}_{3}\right) 1.96$ and 2.14 (each $3 \mathrm{H}, \mathrm{s}, 2$ - and $3-\mathrm{Me}$ ), 3.12br ( $2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}$ ), 3.72 and 3.76 (each $3 \mathrm{H}, \mathrm{s}, 4$ - and $7-\mathrm{OMe}$ ), $6.45(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H})$, and $6.63(1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}$, $\mathrm{Ar}-\mathrm{H}) ; m / e 204\left(M^{+}, 100 \%\right)$ and $189\left(M^{+}-\mathrm{Me}\right)$ (Found: C, 76.2; $\mathrm{H}, 7.8$. $\mathrm{C}_{18} \mathrm{H}_{16} \mathrm{O}_{2}$ requires $\mathrm{C}, 76.4 ; \mathrm{H}, 7.9 \%$ ).
cis- (18) and trans- (19) Dimethoxy-1,3-dimethylisochro-man.-(a) A solution of the indene ( 15 ) ( $1.78 \mathrm{~g}, 9.7 \mathrm{mmol}$ ) and osmium tetraoxide ( $25-50 \mathrm{mg}, 0.1-0.2 \mathrm{mmol}$ ) in dioxan ( 60 ml ) and water $(20 \mathrm{ml})$ was stirred at room temperature for 5 min and then sodium metaperiodate ( $3.91 \mathrm{~g}, 0.018 \mathrm{~mol}$ ) was added, in portions, to the solution over 30 min . The mixture was stirred for 3 days, diluted with water, and extracted with ether. The ethereal extract was washed with water and dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent evaporated off under reduced pressure to give 3-acetonyl-2-acetyl-1,4-dimethoxybenzene (16) (1.48 g, 72\%); $\nu_{\text {max. }} 1720$ and $1700 \mathrm{~cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right) 2.02$ and 2.32 (each $3 \mathrm{H}, \mathrm{s}, 2 \times \mathrm{COMe}$ ), $3.45\left(2 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{2}\right), 3.52$ and 3.56 (each $3 \mathrm{H}, \mathrm{s}, \mathrm{l}$ - and $4-\mathrm{OMe}$ ), and 6.45 ( $2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}$ ) ; $m / e 236$ $\left(M^{+}\right), 221\left(M^{+}-\mathrm{Me}\right), 194\left(M^{+}-\mathrm{CH}_{2}: \mathrm{C}: \mathrm{O}\right)$, and 179 $\left(\mathrm{C}_{10} \mathrm{H}_{11} \mathrm{O}_{3}{ }^{+}, 100 \%\right)$. A mixture of this crude diketone (16) $(0.2 \mathrm{~g}, 0.85 \mathrm{mmol})$ and lithium aluminium hydride $(0.3 \mathrm{~g}$, 7.9 mmol ) in ether ( 20 ml ) was refluxed for 3 h . Saturated Rochelle salt (potassium sodium tartrate) solution was added and the product was extracted with chloroform. Evaporation of the dried $\left(\mathrm{MgSO}_{4}\right)$ organic layer left 2 -(1-hydroxyethyl)-3-(2-hydroxypropyl)-1,4-dimethoxybenzene (17) $(0.172 \mathrm{~g}, 85 \%)$ as an oil; $\nu_{\text {max. }}$ (neat) $\mathbf{3} 400 \mathrm{~cm}^{-1}$. A solution of the crude diol (17) ( $0.172 \mathrm{~g}, 0.72 \mathrm{mmol}$ ) in ether ( 20 ml ) was stirred with concentrated hydrochloric acid ( 2 ml ) at $0^{\circ} \mathrm{C}$ for 1 h , washed with water and dilute sodium hydrogencarbonate solution, and dried $\left(\mathrm{MgSO}_{4}\right)$. The solvent was evaporated off and the residual oil distilled under reduced pressure to afford the 1,3 -dimethylisochromans $(0.115 \mathrm{~g}$, $73 \%$ ), b.p. $90-100^{\circ} \mathrm{C}$ at 1 Torr, showing two spots on t.l.c. owing to the presence of two diastereoisomers, $R_{\mathrm{F}} 0.85$ and $0.80\left(\mathrm{SiO}_{2}\right.$; benzene-ethyl acetate, $\left.5: 1\right)$. The individual isomers were obtained by preparative t.l.c. $\left(\mathrm{SiO}_{2}\right.$; benzeneethyl acetate, $9: 1$ ) as pale yellow oils. The less polar compound was cis-5,8-dimethoxy-1,3-dimethylisochroman (18) ( $26 \mathrm{mg}, 16 \%$ ); $\delta\left(\mathrm{CDCl}_{3}\right) 1.34(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 3-\mathrm{Me}$ ), $1.54(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 1-\mathrm{Me}), 2.34(1 \mathrm{H}$, ddd, $J 17,11$, and $\left.2 \mathrm{~Hz}, 4 \mathrm{a}^{\prime}-\mathrm{H}\right), 2.84\left(1 \mathrm{H}\right.$, ddd, $J 17,2.5$, and $1 \mathrm{~Hz}, 4 \mathrm{e}^{\prime}-$ $\mathrm{H}), 3.50-3.80(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 3.84(6 \mathrm{H}, \mathrm{s}, 5-\mathrm{and} 8$ OMe), 5.10br ( $1 \mathrm{H}, \mathrm{q}, J 7 \mathrm{~Hz}, 1-\mathrm{H}$ ), and $6.80(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H})$; $m / e 222\left(M^{+}\right)$and $207\left(M^{+}-\mathrm{Me}, 100 \%\right.$ ) (Found: $M^{+}$, $222.1248 . \quad \mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{3}$ requires $M, 222.125$ 7). The more polar compound was trans-5,8-dimethoxy-1,3-dimethylisochroman (19) ( $57 \mathrm{mg}, 36 \%$ ); $\delta\left(\mathrm{CDCl}_{3}\right) 1.31(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}$, $3-\mathrm{Me}), 1.49(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 1-\mathrm{Me}), 2.28$ ( $1 \mathrm{H}, \mathrm{dd}, J 17$ and $11 \mathrm{~Hz}, 4 \mathrm{a}^{\prime}-\mathrm{H}$ ), 2.78 ( 1 H , dd, $J 17$ and $3 \mathrm{~Hz}, 4 \mathrm{e}^{\prime}-\mathrm{H}$ ), 3.76 $(6 \mathrm{H}, \mathrm{s}, 5-\mathrm{and} 8-\mathrm{OMe}), 3.90-4.20(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 5.10(1 \mathrm{H}$, $\mathrm{q}, J 6 \mathrm{~Hz}, \mathrm{l}-\mathrm{H})$, and $6.64(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}) ; m / e 222\left(M^{+}\right)$and $207\left(M^{+}-\mathrm{Me}, 100 \%\right)$ (Found: $M^{+}$, 222.129 2. $\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{3}$ requires $M, 222.1257$ ). The ratio of cis- to trans-isomers obtained here was determined by g.l.c. to be $1: 2\left(t_{R} 4.7\right.$ and $5.7 \mathrm{~min} ; 20 \% \mathrm{SE}-30$ on Shimalite W at $160^{\circ} \mathrm{C}$ ). The cis-trans ratios obtained by using other catalysts at room temperature for 10 min were as follows: $1: 3$ with $1 \%$ $\mathrm{H}_{2} \mathrm{SO}_{4}$ in AcOH ( $80 \%$ yield); $1: 4$ with $\mathrm{MeSO}_{3} \mathrm{H}$ ( $70 \%$ yield).
(b) A solution of hydrogen bromide in acetic acid ( $26 \%$; $3 \mathrm{ml})$ was added to a solution of the diketone (16) $(0.714 \mathrm{~g}$,

3 mmol ) in acetic acid ( 2 ml ). After a few minutes, ethyl acetate ( 50 ml ) was added to the mixture and filtration of the resulting reddish precipitate gave 5,8-dimethoxy-1,3dimethylisochromenylium bromide (21) (0.578 g, 64\%); $\nu_{\max }(\mathrm{KBr}) 1640,1600,1560$, and $1490 \mathrm{~cm}^{-1}, \delta\left(\mathrm{CD}_{3} \mathrm{CO}_{2} \mathrm{D}\right)$ 2.07 and 2.94 (each $3 \mathrm{H}, \mathrm{s}, 1$ - and $3-\mathrm{Me}$ ), 4.17 and 4.29 (each $3 \mathrm{H}, \mathrm{s}, 5$ - and 8 -OMe), 7.64 ( $1 \mathrm{H}, \mathrm{s}, J 8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}$ ), 8.16 ( $1 \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, \mathrm{Ar}-\mathrm{H}$ ), and 8.36 ( $1 \mathrm{H}, \mathrm{s}, 4-\mathrm{H}$ ) ; $m / e 219$ $\left(M^{+}-\mathrm{Br}\right), 218\left(M^{+}-\mathrm{HBr}\right)$, and $213\left(M^{+}-\mathrm{HBr}-\mathrm{Me}\right.$, $100 \%)$. The oxonium salt (21) ( $0.2 \mathrm{~g}, 0.7 \mathrm{mmol}$ ) was catalytically hydrogenated in acetic acid ( 6 ml ) over $\mathrm{PtO}_{2}(0.1 \mathrm{~g})$ under atmospheric pressure for 24 h . After the catalyst and the solvent were removed, the residual oil was dissolved in ether, washed with dilute sodium hydrogencarbonate solution and water, dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent removed under reduced pressure to afford an oil ( 91 mg , $60 \%$ ), which was identical (n.m.r. spectrum) with cis-5,8-dimethoxy-1,3-dimethylisochroman (18).

Epimerisation of cis-5,8-Dimethoxy-1,3-dimethylisochro$\operatorname{man}$ (18).-A solution of the cis-isochroman (18) ( 10 mg , 0.045 mmol ) in methanesulphonic acid ( 1 ml ) was stirred at room temperature for 10 min . The resulting mixture was diluted with water, extracted with ethyl acetate, the extract dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent removed under reduced pressure to give an oil ( $8 \mathrm{mg}, 80 \%$ ), which was a $c a$ 1:4 mixture of the starting cis-isochroman (18) and the trans-isomer (19) (by g.l.c.).
cis- (22) and trans- (23) 1,3-Dimethylisochroman-5,8-dione.-A solution of cerium(IV) ammonium nitrate ( 0.532 $\mathrm{g}, 0.97 \mathrm{mmol}$ ) in water ( 0.8 ml ) was added dropwise, over 5 min , to a stirred solution of the cis-isochroman (18) $(72 \mathrm{mg}$, 0.324 mmol ) in acetonitrile ( 0.8 ml ) at room temperature and the mixture was stirred for 30 min , diluted with water and extracted with chloroform. The extract was washed with water, dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent evaporated off. The residual oil was distilled under reduced pressure to afford cis-1,3-dimethylisochroman-5,8-dione (22) $(53 \mathrm{mg}$, $85 \%$ ) as a yellow oil, b.p. $100-110^{\circ} \mathrm{C}$ at 1 Torr; $v_{\max }$ (neat) $1655 \mathrm{~cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right) 1.32(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 3-\mathrm{Me}), 1.44$ $(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 1-\mathrm{Me}), 2.08(1 \mathrm{H}$, ddd, $J 18,11$, and 4 Hz , $4-\mathrm{H}), 2.59(1 \mathrm{H}, \mathrm{dt}, J 18$ and $2.8 \mathrm{~Hz}, 4-\mathrm{H}), 3.40-3.60(1 \mathrm{H}$, $\mathrm{m}, 3-\mathrm{H}), 4.45-4.65(1 \mathrm{H}, \mathrm{m}, \mathrm{l}-\mathrm{H})$, and 6.62 ( $2 \mathrm{H}, \mathrm{s}$, $\mathrm{Ar}-\mathrm{H}) ; \quad m / e \quad 192\left(M^{+}\right)$and $177 \quad\left(M^{+}-\mathrm{Me}, \quad 100 \%\right)$ (Found: C, 68.5; H, 6.1. $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{O}_{3}$ requires $\mathrm{C}, 68.7$; H , $6.3 \%)$.
Similarly, the trans-isochroman (19) ( $56 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) was converted into the trans-isomer (23) ( $41 \mathrm{mg}, 85 \%$ ) m.p. $102.5-105.5{ }^{\circ} \mathrm{C}$ (from EtOH ); $\nu_{\text {max }}(\mathrm{KBr}) 1660$ $\mathrm{cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right) 1.33(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 3-\mathrm{Me}), 1.45(3 \mathrm{H}, \mathrm{d}, J$ $6 \mathrm{~Hz}, 1-\mathrm{Me}$ ), 2.08 ( 1 H , ddd, $J 19,10$, and $2 \mathrm{~Hz}, 4-\mathrm{H}$ ), $2.56(1 \mathrm{H}, \mathrm{dd}, J 19$ and $4 \mathrm{~Hz}, 4-\mathrm{H}), 3.70-4.10(1 \mathrm{H}, \mathrm{m}$, $3-\mathrm{H}), 4.80 \mathrm{br}(1 \mathrm{H}, \mathrm{q}, J 6 \mathrm{~Hz}, 1-\mathrm{H})$, and $6.66(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}$ H) ; $m / e 192\left(M^{+}\right)$and $177\left(M^{+}-\mathrm{Me}, 100 \%\right)$ (Found: C, 69.0 ; H, 6.2\%).
cis- (22) and trans- (iv)(22) 1,3-Dimethyl-3,4-dihydro-1H-naphtho[2,3-c]pyran-5,10-dione.-A solution of the cisquinone (22) ( $50 \mathrm{mg}, 0.26 \mathrm{mmol}$ ) and l-acetoxybutadiene $(0.1 \mathrm{~g}, 0.89 \mathrm{mmol})$ in toluene ( 1 ml ) was set aside at room temperature for 3 days. The mixture was evaporated to dryness under reduced pressure and the residual oil was dissolved in ethanol ( 5 ml ). To the solution was added $1 \%$ sodium carbonate solution ( 0.5 ml ) and the mixture was stirred at room temperature for 5 h , diluted with ethyl acetate, and washed with water. Evaporation of the dried $\left(\mathrm{MgSO}_{4}\right)$ organic layer left a yellow solid, which was re-
crystallised from ethanol to afford cis-1,3-dimethyl-3,4-dihydro-1H-naphtho[2,3-c]pyran-5,10-dione (24) $(56 \mathrm{mg}$, $89 \%$ ), m.p. $122-125{ }^{\circ} \mathrm{C}$; $v_{\max }(\mathrm{KBr}) 1660 \mathrm{~cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right)$ $1.39(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 3-\mathrm{Me})$, $1.56(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 1-\mathrm{Me})$, 2.24 ( 1 H , ddd, $J 18,11$, and $3.8 \mathrm{~Hz}, 4-\mathrm{H}$ ), $2.80(1 \mathrm{H}, \mathrm{dt}$, $J 18$ and $2.7 \mathrm{~Hz}, 4-\mathrm{H}), 3.50-3.80(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 4.70-5.00$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{l}-\mathrm{H}$ ), $7.60-7.80(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H})$, and $7.95-8.20$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ) ; m/e $242\left(M^{+}, 100 \%\right)$ and $227\left(M^{+}-\mathrm{Me}\right)$ (Found: $\mathrm{C}, 74.55 ; \mathrm{H}, 5.7 . \quad \mathrm{C}_{15} \mathrm{H}_{14} \mathrm{O}_{3}$ requires $\mathrm{C}, 74.4 ; \mathrm{H}$, $5.8 \%$ ).

Similarly, the trans-quinone (23) ( $56 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) was converted into the trans-isomer (25) ( $55 \mathrm{mg}, 91 \%$ ), m.p. $146-148{ }^{\circ} \mathrm{C}$ (from EtOH); $\nu_{\text {max. }}(\mathrm{KBr}) 1660 \mathrm{~cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right)$ $1.40(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 3-\mathrm{Me})$, $1.57(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 1-\mathrm{Me})$, 2.18 ( 1 H , ddd, $J 18,10$, and $2 \mathrm{~Hz}, 4-\mathrm{H}$ ), $2.80(1 \mathrm{H}$, dd, $J 18$ and $3.2 \mathrm{~Hz}, 4-\mathrm{H}), 3.90-4.20(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 5.60 \mathrm{br}(1 \mathrm{H}, \mathrm{q}$, $J 6 \mathrm{~Hz}, 1-\mathrm{H}), 7.70-7.90(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H})$, and $8.05-8.25$ ( $2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ) ; m/e $242\left(M^{+}, 100 \%\right.$ ) and $227\left(M^{+}-\mathrm{Me}\right)$ (Found: C, 74.2; H, 5.8\%).

4,7-Dimethoxy-3-methylindene (32).-An ethereal solution of methylmagnesium iodide ( 2 m -solution; $1 \mathrm{ml}, 2 \mathrm{mmol}$ ) was added dropwise, over 5 min , to a stirred solution of 4,7-dimethoxyindan-1-one ${ }^{13}$ ( 28 ) ( $0.2 \mathrm{~g}, 1.04 \mathrm{mmol}$ ) in ether $(15 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$ and the mixture was stirred at room temperature for 1 h , poured into saturated ammonium chloride solution, and extracted with ether. The ethereal extract was vigorously shaken with $10 \%$ hydrochloric acid for 2 min , washed with dilute sodium hydrogencarbonate solution and dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent removed under reduced pressure. Column chromatography on silica gel, with benzene as eluant, afforded 4,7-dimethoxy-3-methylindene (32) ( $0.158 \mathrm{~g}, 80 \%$ ) as a pale yellow solid, m.p. $50-55$ ${ }^{\circ} \mathrm{C}$; $\delta\left(\mathrm{CDCl}_{3}\right) 2.26(3 \mathrm{H}, \mathrm{q}, J 2 \mathrm{~Hz}, 3-\mathrm{Me}), 3.14(2 \mathrm{H}$, quintet, $J 2 \mathrm{~Hz}, \mathrm{CH}_{2}$ ), 3.72 and 3.75 (each $3 \mathrm{H}, \mathrm{s}, 4$ - and $7-\mathrm{OMe}$ ), 5.93 ( 1 H , sextet, $J 2 \mathrm{~Hz}, 2-\mathrm{H}$ ), and $6.44-6.70(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H})$; $m / e 190\left(M^{+}, 100 \%\right)$ and $175\left(M^{+}-\mathrm{Me}\right)$.

2-Acetyl-3-(3-methoxycarbonylprop-2-enyl)-1,4-dimethoxybenzene (34).-A solution of the indene (32) ( $0.3 \mathrm{~g}, 1.58$ mmol ) and osmium tetraoxide ( $25-50 \mathrm{mg}, 0.1-0.2 \mathrm{mmol}$ ) in dimethoxyethane ( 6 ml ) and water ( 2 ml ) was stirred at room temperature for 5 min and then sodium metaperiodate ( $0.705 \mathrm{~g}, 3.3 \mathrm{mmol}$ ) was added, in portions, to the solution over 30 min . The mixture was stirred for 4 h , diluted with water, and extracted with ether. The ethereal extract was washed with water and dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent evaporated off under reduced pressure to give 2 -acetyl3 -formylmethyl-1,4-dimethoxybenzene (33) (0.278 g, 79\%) as a pale yellow oil; $\nu_{\text {max }}$ (neat) 2720,1725 , and 1690 $\mathrm{cm}^{-1}$; $\delta\left(\mathrm{CDCl}_{3}\right) 2.44(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.62(2 \mathrm{H}, \mathrm{d}, J 1 \mathrm{~Hz}$, $\mathrm{CH}_{2}$ ), 3.72 and 3.75 (each $3 \mathrm{H}, \mathrm{s}, 1$ - and 4-OMe), 6.77 ( $2 \mathrm{H} . \mathrm{s}$. $\mathrm{Ar}-\mathrm{H})$, and $9.56(1 \mathrm{H}, \mathrm{t}, J 1 \mathrm{~Hz}, \mathrm{CHO}) ; m / e 222\left(M^{+}\right)$, $194\left(M^{+}-\mathrm{CO}\right)$, and $179\left(M^{+}-\mathrm{CO}-\mathrm{Me}, 100 \%\right)$. A solution of this crude keto-aldehyde (33) ( $0.278 \mathrm{~g}, 1.25$ mmol ) and methoxycarbonylmethylenetriphenylphosphorane ( $0.5 \mathrm{~g}, 1.5 \mathrm{mmol}$ ) in benzene ( 12 ml ) was stirred at room temperature for 24 h . The resulting solution was concentrated under reduced pressure and the residue was chromatographed on silica gel, with benzene as eluant, to yield 2-acetyl-3-(3-methoxycarbonylprop-2-enyl)-1,4-dimethoxybenzene (34) ( $0.115 \mathrm{~g}, 45 \%$ ) as an oil, b.p. $160-170{ }^{\circ} \mathrm{C}$ at 1 Torr; $\delta\left(\mathrm{CDCl}_{3}\right) 2.38(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.40(2 \mathrm{H}, \mathrm{dd}, J 7$ and $1.5 \mathrm{~Hz}, \mathrm{CH}_{2}$ ), $3.64(3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}), 3.80(6 \mathrm{H}, \mathrm{s}, 2 \times$ $\mathrm{OMe}), 5.60\left(1 \mathrm{H}, \mathrm{dt}, J 16\right.$ and $\left.1.5 \mathrm{~Hz}, \mathrm{C}: \mathrm{CH} \cdot \mathrm{CO}_{2} \mathrm{Me}\right)$, $6.75(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H})$, and $6.90(1 \mathrm{H}, \mathrm{dt}, J 16$ and 7 Hz , $\left.\mathrm{CH}: \mathrm{CH} \cdot \mathrm{CO}_{2} \mathrm{Me}\right) ; \quad m / e 278\left(M^{+}\right), 263\left(\mathrm{C}_{13} \mathrm{H}_{18} \mathrm{O}_{4}{ }^{+}\right), 205$
$\left(\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{O}_{3}{ }^{+}, 100 \%\right)$, and $204\left(\mathrm{C}_{12} \mathrm{H}_{13} \mathrm{O}_{3}{ }^{+}\right)$(Found: $\mathrm{C}, 64.8$; $\mathrm{H}, 6.3$. $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{O}_{5}$ requires $\mathrm{C}, 64.7 ; \mathrm{H}, 6.5 \%$ ).
cis- (35) and trans- (36) Methyl 5,8-Dimethoxy-1-methyl-isochroman-3-ylacetate.-Sodium borohydride ( $0.5 \mathrm{~g}, 0.013$ $\mathrm{mol})$ was added at $0{ }^{\circ} \mathrm{C}$, in portions, to a stirred solution of the conjugated ester (34) ( $0.15 \mathrm{~g}, 0.54 \mathrm{mmol}$ ) in methanol $(10 \mathrm{ml})$ over 30 min and the mixture was stirred at room temperature for 24 h , poured into water, and extracted with ethyl acetate. The extract was washed with water, dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent evaporated off under reduced pressure to give isochromans ( $0.138 \mathrm{~g}, 92 \%$ ) as a pale yellow oil showing two spots on t.l.c. owing to the presence of two diastereoisomers, $R_{\mathrm{F}} 0.55$ and 0.50 ( $\mathrm{SiO}_{2}$; benzene-ethyl acetate, $10: 1$ ). The individual isomers were obtained by preparative t.l.c. $\left(\mathrm{SiO}_{2}\right.$; benzene-ethyl acetate, $9: 1$ ) as pale yellow oils. The less polar conipound was methyl cis-5,8-dimethoxy-1-methylisochroman-3-ylacetate. (35) ( $26 \mathrm{mg}, 17 \%$ ); ${ }^{{ }_{\text {max. }}}$ (neat) $1735 \mathrm{~cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right) 1.46$ $(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 1-\mathrm{Me}), 2.18(1 \mathrm{H}, \mathrm{ddd}, J 17,11$, and 1.5 $\mathrm{Hz}, 4-\mathrm{H}), 2.62\left(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 6 \mathrm{~Hz}, \mathrm{C} \mathrm{H}_{2} \cdot \mathrm{CO}_{2} \mathrm{Me}\right), 2.80(1 \mathrm{H}$, ddd, $J$ 17, 2.7, and $1.5 \mathrm{~Hz}, 4-\mathrm{H}$ ), 3.63 ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), 3.60 ( $6 \mathrm{H}, \mathrm{s}$, two OMe ), $3.70-4.00(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 4.91(1 \mathrm{H}, \mathrm{q}$, $J 6 \mathrm{~Hz}, 1-\mathrm{H})$, and $6.52(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}) ; m / e 280\left(M^{+}, 100 \%\right)$, $265\left(M^{+}-\mathrm{Me}\right)$, and $262\left(M^{+}-\mathrm{H}_{2} \mathrm{O}\right)$ (Found: $M^{+}$, $280.1315 . \quad \mathrm{C}_{15} \mathrm{H}_{20} \mathrm{O}_{5}$ requires $M, 280.131$ 2). The more polar compound was the trans-isomer (36) ( $60 \mathrm{mg}, \mathbf{4 0 \%}$ ); $\nu_{\text {max. }}$ (neat) $1740 \mathrm{~cm}^{-1} ; \delta\left(\mathrm{CDCl}_{3}\right) 1.45(3 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 1-\mathrm{Me})$, $2.27(1 \mathrm{H}, \mathrm{dd}, J 17$ and $11 \mathrm{~Hz}, 4-\mathrm{H}), 2.53(2 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}$, $\mathrm{CH}_{2} \cdot \mathrm{CO}_{2} \mathrm{Me}$ ), $2.73(1 \mathrm{H}, \mathrm{dd}, J 17$ and $3.5 \mathrm{~Hz}, 4-\mathrm{H}), 3.63$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{OMe}$ ), 3.65 ( 6 H , s, two OMe), $4.10-4.40$ ( $1 \mathrm{H}, \mathrm{m}$, $3-\mathrm{Me}), 4.90(1 \mathrm{H}, \mathrm{q}, J 6 \mathrm{~Hz}, \mathrm{l}-\mathrm{H})$, and $6.49(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H})$; $m / e 280\left(M^{+}, 100 \%\right), 265\left(M^{+}-\mathrm{Me}\right)$, and $262\left(M^{+}-\right.$ $\mathrm{H}_{2} \mathrm{O}$ ) (Found: $M^{+}, 280.1328$ ). The cis- to trans-isomer ratio was determined by g.l.c. to be 1:3.5 ( $t_{1} 2.7$ and 3.3 $\min ; 20 \%$ SE- 30 on Shimalite W at $220^{\circ} \mathrm{C}$ ).
cis- (37) and trans- (38) Methyl 1-Methyl-5,8-dioxo-5,8-dihydroisochroman-3-ylacetate.-A solution of cerium(iv) ammonium nitrate ( $0.49 \mathrm{~g}, 0.9 \mathrm{mmol}$ ) in water ( 1 ml ) was added dropwise, over 5 min , to a stirred solution of the cisisochroman (35) ( $85 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) in acetonitrile ( 1 ml ) at room temperature. The mixture was stirred for 30 min, diluted with water, and extracted with ethyl acetate. The extract was washed with water, dried $\left(\mathrm{MgSO}_{4}\right)$, and the solvent evaporated off. The resulting oil was chromatographed on silica gel, with benzene as eluant, afforded methyl cis-1-methyl-5,8-dioxo-5,8-dihydroisochroman-3-ylacetate (37) ( $65 \mathrm{mg}, 86 \%$ ) as a dark yellow oil; ${ }^{\nu_{\text {max. }}}$ (neat, 1740 and $1660 \mathrm{~cm}^{-1}$; $\delta\left(\mathrm{CDCl}_{3}\right) 1.44(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 1-\mathrm{Me})$, 2.17 ( 1 H, ddd, $J 18,11$, and $3.7 \mathrm{~Hz}, 4-\mathrm{H}$ ), $2.62(2 \mathrm{H}, \mathrm{dd})$ $J 6$ and $\left.2.9 \mathrm{~Hz}, \mathrm{CH}_{2} \cdot \mathrm{CO}_{2} \mathrm{Me}\right), 2.66(1 \mathrm{H}, \mathrm{dt}, J 18$ and 2.5 $\mathrm{Hz}, 4-\mathrm{H}), 3.68\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{Me}\right), 3.70-4.00(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$, $4.50-4.80(1 \mathrm{H}, \mathrm{m}, \mathrm{l}-\mathrm{H})$, and $6.67(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}) ; m / e$ $250\left(M^{+}\right), 235\left(M^{+}-\mathrm{Me}\right), 218\left(M^{+}-\mathrm{MeOH}\right), 177\left(\mathrm{C}_{10^{-}}\right.$ $\mathrm{H}_{8} \mathrm{O}_{3}{ }^{+}$), and $176\left(\mathrm{C}_{10} \mathrm{H}_{8} \mathrm{O}_{3}{ }^{+}\right)$.

Similarly, the trans-isochroman (36) ( $0.139 \mathrm{~g}, 0.5 \mathrm{mmol}$ ) was converted into the trans-isomer (38) ( $95 \mathrm{mg}, 77 \%$ ), m.p. $99-100.5{ }^{\circ} \mathrm{C}$ (from MeOH ); $\nu_{\max .}$ (neat) 1740 and $1660 \mathrm{~cm}^{-1}$; $\delta\left(\mathrm{CDCl}_{3}\right) 1.47(3 \mathrm{H}, \mathrm{d}, J 7 \mathrm{~Hz}, 1-\mathrm{Me}), 2.16(1 \mathrm{H}$,
ddd, $J$ 19, 11, and $2.2 \mathrm{~Hz}, 4-\mathrm{H}), 2.61(2 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}$, $\mathrm{CH}_{2} \cdot \mathrm{CO}_{2} \mathrm{Me}$ ), 2.68 ( 1 H , dd, J 19 and $3.5 \mathrm{~Hz}, 4-\mathrm{H}$ ), 3.73 $\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CO}_{2} \mathrm{Me}\right), 4.20-4.40(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H}), 4.84(1 \mathrm{H}, \mathrm{q}$, $J 7 \mathrm{~Hz}, 1-\mathrm{H})$, and $6.75(2 \mathrm{H}, \mathrm{s}, \mathrm{Ar}-\mathrm{H}) ; m / e 250\left(M^{+}\right), 235$ $\left(M^{+}-\mathrm{Me}\right), 218\left(M^{+}-\mathrm{MeOH}\right), 177\left(\mathrm{C}_{10} \mathrm{H}_{9} \mathrm{O}_{3}{ }^{+}, 100 \%\right)$, and $176\left(\mathrm{C}_{10} \mathrm{H}_{8} \mathrm{O}_{3}{ }^{+}\right)$(Found: C, 62.4; H, 5.6. $\mathrm{C}_{13} \mathrm{H}_{14} \mathrm{O}_{5}$ requires $\mathrm{C}, 62.4 ; \mathrm{H}, 5.6 \%$ ).

Methyl trans-1-Methyl-5,10-dioxo-3,4,5,10-tetrahydro-1H-naphtho[2,3-c]pyran-3-ylacetate (39).-A solution of the trans-quinone (38) ( $30 \mathrm{mg}, 0.12 \mathrm{mmol}$ ) and l-acetoxybutadiene ( $60 \mathrm{mg}, 0.54 \mathrm{mmol}$ ) in toluene ( 0.7 ml ) was set aside at room temperature for 2 days. The mixture was evaporated to dryness under reduced pressure and the residual oil was dissolved in ethanol ( 3 ml ). To the solution was added $1 \%$ sodium carbonate solution ( 0.3 ml ) and the mixture was stirred at room temperature for 5 h , diluted with ethyl acetate, and washed with water. Evaporation of the dried $\left(\mathrm{MgSO}_{4}\right)$ organic layer left a yellow solicl, which was recrystallised from methanol to afford methyl trans-1-methyl-5,10-dioxo-3,4,5,10-tetrahydro-1 $/ I$-naphtho [2,3-
c]pyran-3-ylacetate (39) ( $28 \mathrm{mg}, 78 \%$ ) as yellow needles, m.p. $185-187^{\circ} \mathrm{C}$ (lit., ${ }^{6}$ m.p. $185-187^{\circ} \mathrm{C}$ ); $v_{\text {дах. }}\left(\mathrm{KBr}^{\prime}\right)$ 1735 and $1660 \mathrm{~cm}^{-1}$, the n.m.r. clata of which were completely identical with those reported by l'yrek, ${ }^{6} \mathrm{~m} / \mathrm{e} 300$ $\left(M^{+}\right), 268\left(M^{+}-\mathrm{MeOH}\right), 227\left(\mathrm{C}_{14} \mathrm{H}_{11} \mathrm{O}_{3}{ }^{-1}\right)$, and $226\left(\mathrm{C}_{144^{-}}\right.$ $\mathrm{H}_{10} \mathrm{O}_{3}{ }^{+}, 100 \%$ ) (Found: C, $\mathbf{6 7 . 8}$; $\mathrm{H}, 5.4$. Calc. for $\mathrm{C}_{17} \mathrm{H}_{16}{ }^{-}$ $\left.\mathrm{O}_{5}: \mathrm{H}, 68.0 ; \mathrm{H}, 5.4 \%\right)$.
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